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# LATTICE BOLTZMANN METHOD for CFD

More @ LBE

Dr. Jacques C. Richard, [richard@aero.tamu.edu](mailto:richard@aero.tamu.edu)

Dr. Sharath Girimaji, [girimaji@aero.tamu.edu](mailto:girimaji@aero.tamu.edu)

Dr. Dazhi Yu, [dzyu@aero.tamu.edu](mailto:dzyu@aero.tamu.edu)

Dr. Huidan Yu, [h0y5840@aero.tamu.edu](mailto:h0y5840@aero.tamu.edu)

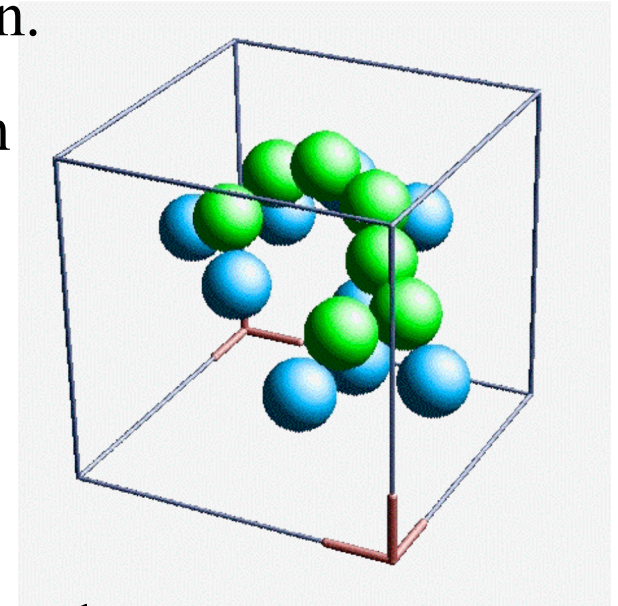
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# What makes up fluids

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- Fluid made of small, individual molecules.
- Molecules are in a state of constant motion.
- Molecules are at continuous collision with each other.
- Molecule has internal structure.
- There are intermolecular forces.
- Macroscopic variables, such as pressure, temperature, and internal energy, are determined by the mass, velocity, and internal structure of molecules.



# Modeling Fluids

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Knudsen number  $Kn \equiv \frac{\text{Mean free path}}{\text{Characteristic hydrodynamic length}}$

👉 Microscopic Method : Molecular dynamics

$$Kn < \infty$$

👉 Mesoscopic Method : Boltzmann Equation

$$Kn < \infty$$

👉 Macroscopic Method: Navier-Stokes Equations

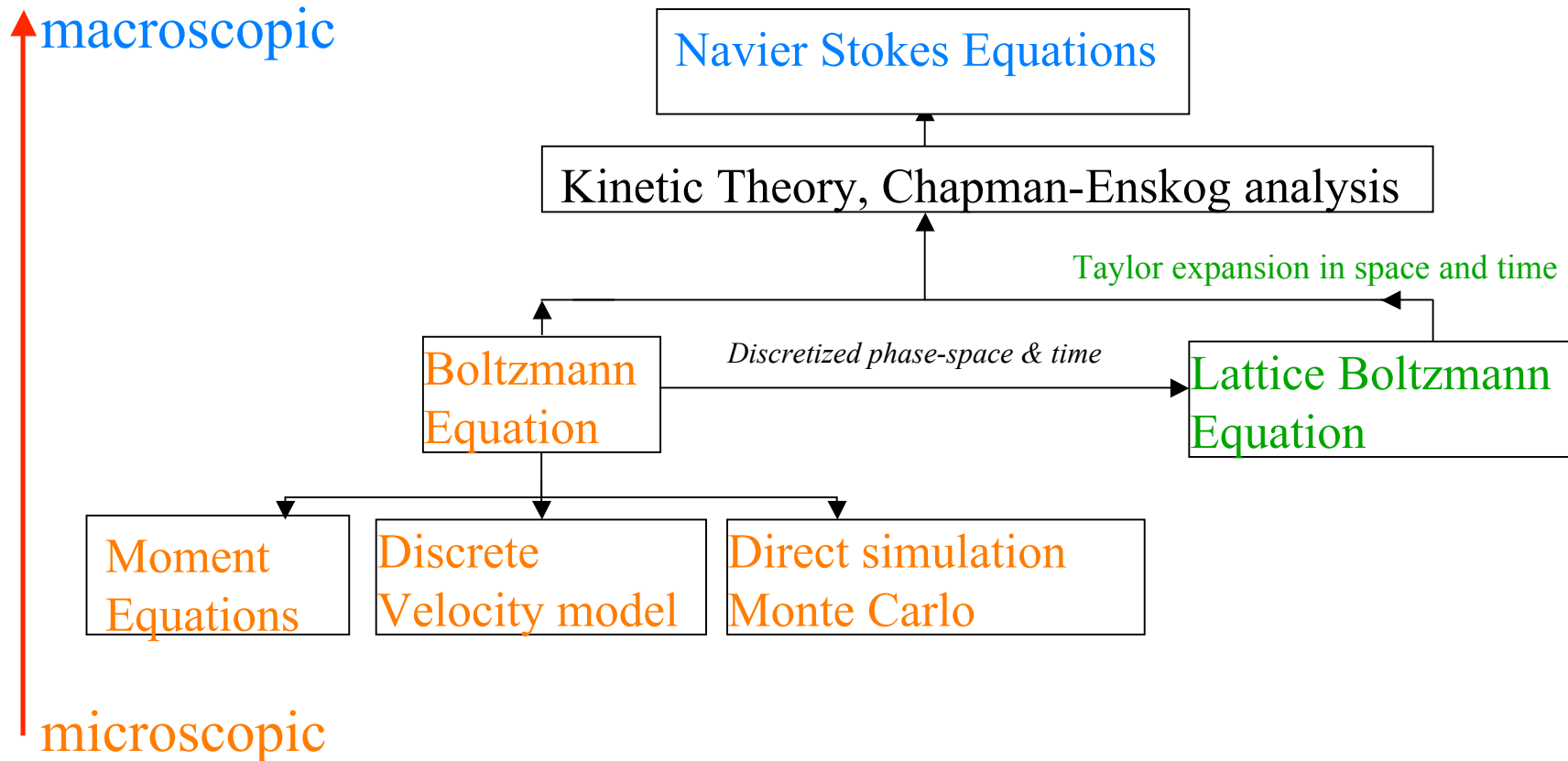
$$Kn < 0.1$$

👉 A Novel Method : Lattice Boltzmann Method

$$Kn < 0.1$$

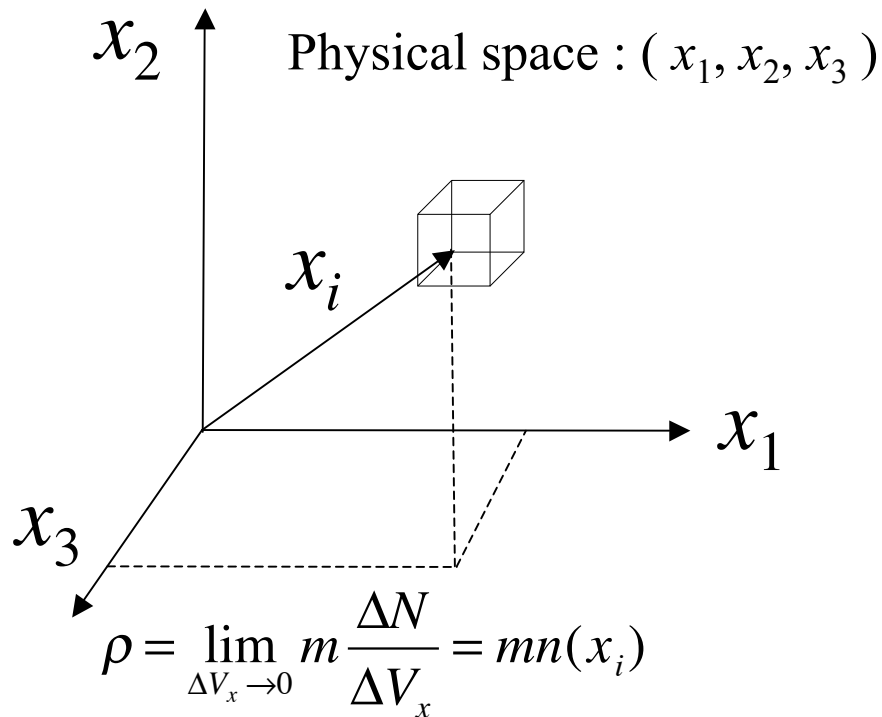


# Modeling of Fluids



# Velocity distribution function

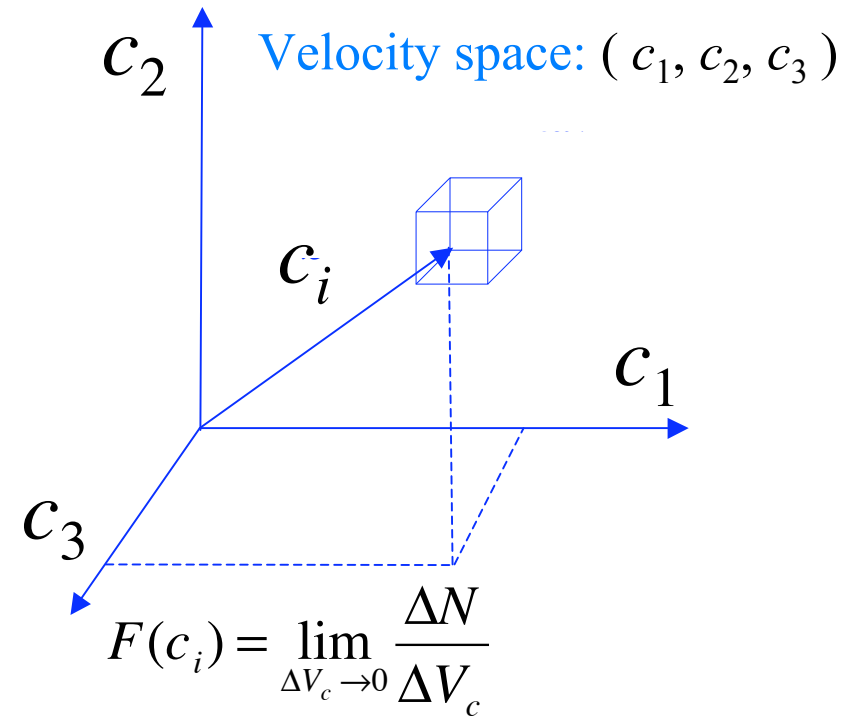
- Velocity distribution function:  $F(c_i)$



$m$  : mass of molecule

$\Delta N$  : number of molecules in  $\Delta V_x$

$n(x_i)$  : local number density



# Velocity Distribution Function

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- Normalized Velocity distribution Function

$$f(c_i) = F(c_i) / N$$

$N$ : Number of molecules in system

- Characteristics of Distribution Function  $f$

- $$\int_{-\infty}^{\infty} Nf(c_i)dV_c = N \Rightarrow \int_{-\infty}^{\infty} f(c_i)dV_c = 1$$

- $$\bar{Q} = \frac{\int_N QdN}{N} = \frac{\int_{-\infty}^{\infty} Q(c_i)Nf(c_i)dV_c}{N} = \int_{-\infty}^{\infty} QfdV_c$$

$\bar{Q}$  is the average value of  $Q$  for all molecules

Macroscopic velocity:  $\vec{u} = \mathbf{u} = \bar{\mathbf{c}} = \int_{-\infty}^{\infty} c_i f dV_c$



# Velocity Distribution Function under Equilibrium State

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- Equilibrium State

In equilibrium state, the number of molecules in  $c_i$  is constant.

- Distribution function under equilibrium state is the Maxwellian distribution:

$$f^{eq}(c_i) = \left( \frac{m}{2\pi kT} \right)^{3/2} e^{-\frac{m}{2kT} [(c_1 - u_1)^2 + (c_2 - u_2)^2 + (c_3 - u_3)^2]}$$

$T$ : temperature;

$k$ : Boltzmann constant ( $1.38054 \times 10^{-16}$  erg-K)

$m$ : mass of molecule

$u$ : macroscopic velocity



# Boltzmann Equation

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- Under non-equilibrium state distribution function

$$f = f(x_i, c_i, t) = f(\mathbf{x}, \mathbf{c}, t)$$

- The molecular velocity distribution function has a rate of change, with respect to position and time, that is described by

$$\frac{\partial}{\partial t} [nf(c_i)] + c_j \frac{\partial}{\partial x_j} [nf(c_i)] = \left\{ \frac{\partial}{\partial t} [nf(c_i)] \right\}_{collision}$$

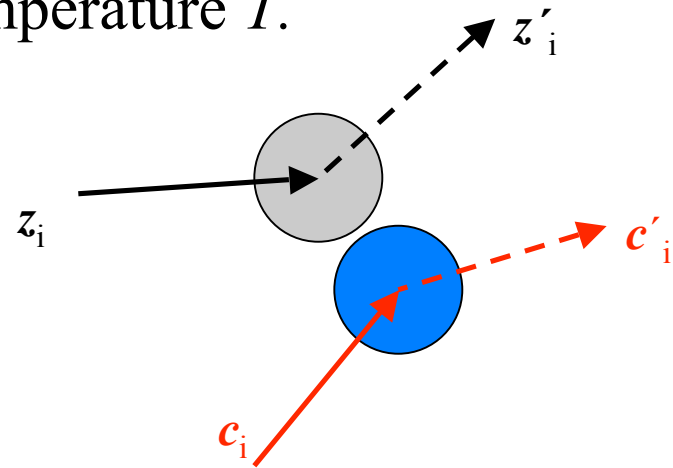




# Collisions in fluids

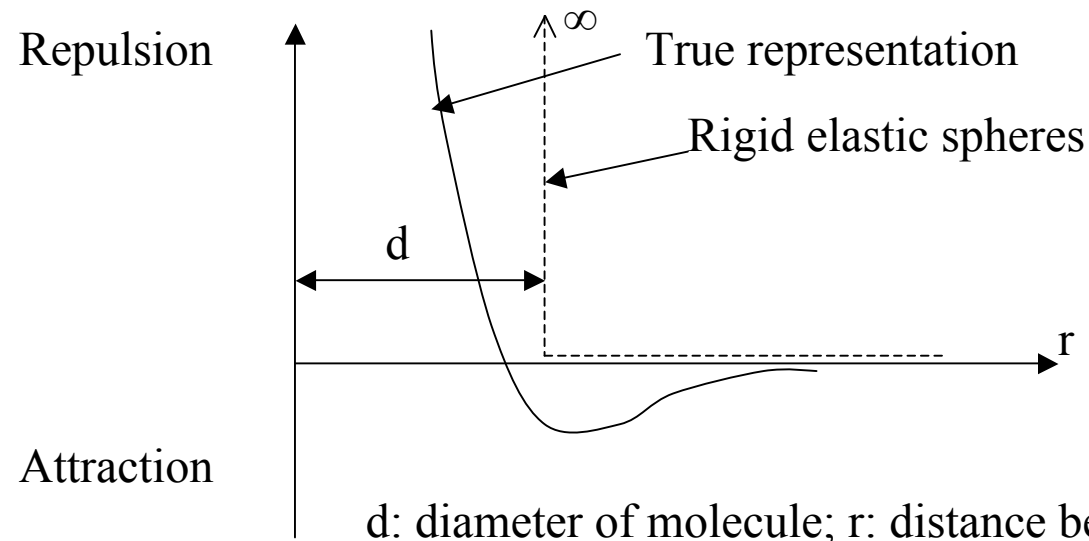
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- Molecules constantly collide with each other.
- Collision will change the velocity of molecules.
- In collision, the translational energy of molecule may transfer to internal energy, if molecule has internal structure.
- In general, the rate of collision depends on the molecule velocity, number density  $n(\mathbf{x})$ , and temperature  $T$ .



# Collision Operator in Boltzmann Equation

- Elastic-sphere molecule model
  1. no internal structure of molecule,
  2. the molecule is treated as a rigid ball,
  3. so no rotation and vibration,
  4. only translation.
- no attraction force between molecules.



# Collision operator

- Rigid ball model gives:

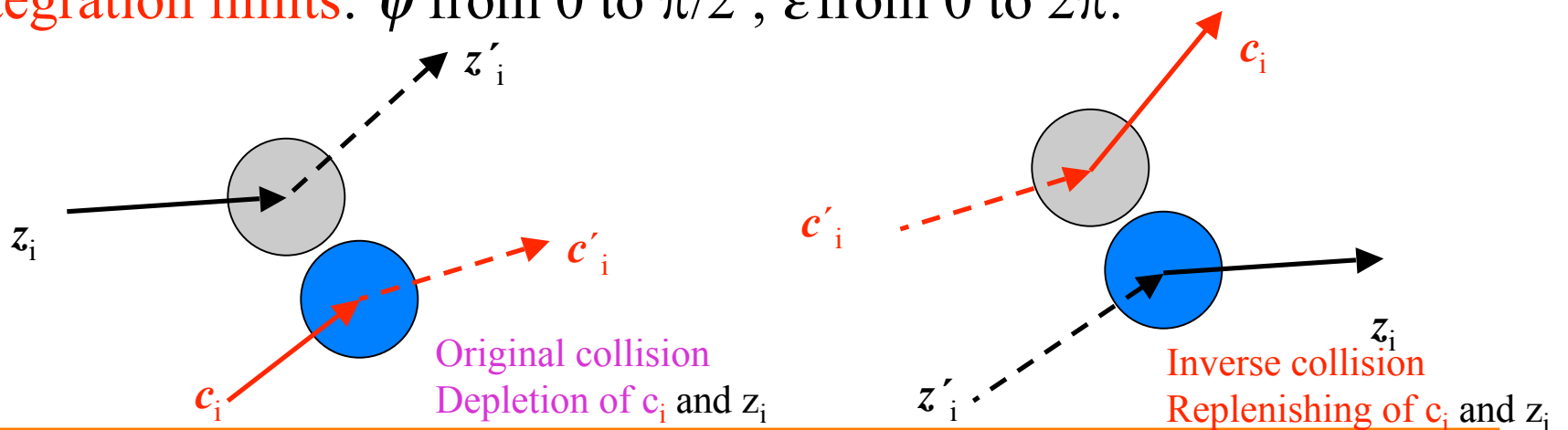
$$\left\{ \frac{\partial}{\partial t} [nf(c_i)] \right\}_{collision} = \int_{-\infty}^{\infty} \int_0^{2\pi} \int_0^{\pi/2} n^2 d^2 [f(c'_i)f(z'_i) - f(c_i)f(z_i)] g \sin \psi \cos \psi d\psi d\varepsilon dV_x$$

$g$ : relative velocity between two collided molecules

$f(c'_i)f(z'_i)$ : replenishing of the molecules of class  $c_i$

$f(c_i)f(z_i)$ : depleting of the molecules of class  $c_i$

Integration limits:  $\psi$  from 0 to  $\pi/2$ ;  $\varepsilon$  from 0 to  $2\pi$ .



# Macroscopic variables

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## Quantities of interest

- Density

$$\rho = \int_{-\infty}^{\infty} m f dV_c$$

- Momentum

$$\rho \mathbf{u} = \int_{-\infty}^{\infty} m \mathbf{c} f dV_c$$

- Translational energy

$$e_{tr} = \int_{-\infty}^{\infty} \frac{1}{2} (\mathbf{c} - \bar{\mathbf{c}})^2 f dV_c$$



# Moments of the Boltzmann Equation

- Let  $Q(c_i)$  be a function of  $c_i$  but not of position and time, the equation of transfer of  $Q(c_i)$  is

$$\int_{-\infty}^{\infty} Q(c_i) \left\{ \frac{\partial}{\partial t} [nf(c_i)] + c_j \frac{\partial}{\partial x_j} [nf(c_i)] \right\} dV_c = \int_{-\infty}^{\infty} Q(c_i) \left\{ \frac{\partial}{\partial t} [nf(c_i)] \right\}_{collision} dV_c$$



- $$\frac{\partial}{\partial t} [n\bar{Q}] + \frac{\partial}{\partial x_j} [n\bar{Q}c_i] = \Delta[Q]$$

if  $Q = m, mc_i, mc^2/2$ , the change in  $Q$  for both molecules must be zero in collision. We have

$$\frac{\partial}{\partial t} [n\bar{Q}] + \frac{\partial}{\partial x_j} [n\bar{Q}c_i] = 0$$

Let  $Q = m$ , we have the continuity equation in NS. We recover the momentum and energy equations in NS when  $Q = mc_i, mc^2/2$



# The Conservation Equations from the Boltzmann Equation

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$$Q = m \quad \frac{\partial \rho}{\partial t} + \frac{\partial}{\partial x_j} (\rho \bar{c}_j) = 0$$

$$Q = mc_i \quad \frac{\partial}{\partial t} (\rho \bar{c}_j) + \frac{\partial}{\partial x_i} (\rho \bar{c}_j \bar{c}_i) = -\frac{\partial p}{\partial x_j} + \frac{\partial \tau_{ij}}{\partial x_i}$$

$$Q = mc^2/2 \quad \frac{\partial}{\partial t} [\rho(e + \bar{c}^2)] + \frac{\partial}{\partial x_i} \left[ \rho \bar{c}_i \left( h + \frac{1}{2} \bar{c}^2 \right) \right] = \frac{\partial}{\partial x_i} (\tau_{ij} \bar{c}_k - q_i)$$

Kinetic theory:  $e = e_{tr} = \overline{C^2}/2$        $h = h_{tr} = e_{tr} + p/\rho = 5RT/2$

$$\rho = nm \quad \overline{p} = \rho [C_1^2 + C_2^2 + C_3^2] / 3$$

New definition:  $\tau_{ij} = -[\rho C_i C_j - p \delta_{ij}]$        $q_j = \rho C_j C^2$

$$\mathbf{C} = \mathbf{c} - \bar{\mathbf{c}}$$

tr: translational



# The Chapmen-Enskog Solution of Boltzmann Equation

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- Non-dimensional form

$$\xi \left\{ \frac{\partial}{\partial \hat{t}} [\hat{n}\hat{f}] + \hat{c}_j \frac{\partial}{\partial \hat{x}_j} [\hat{n}\hat{f}] \right\} = \left\{ \frac{\partial}{\partial \hat{t}} [\hat{n}\hat{f}] \right\}_{\text{collision}}$$

$c_r$ : reference molecule speed ;  $L$ : characteristic length

$\nu_r$ : reference collision frequency

$\xi = c_r / L \nu_r$  is proportional to the Knudsen #, the ratio of the mean free path to a characteristic length, thus it is a very small value.

- Chapman-Enskog expansion (simplifying definitions ( $f = \hat{n}\hat{f}$ ):

$$f(\mathbf{x}, \mathbf{c}, t) = f^{(eq)}(\mathbf{x}, \mathbf{c}, t) + \xi f^{(1)}(\mathbf{x}, \mathbf{c}, t) + \dots$$

Solve the  $f^{(l)}$ , then we get the solution for  $f$



# Chapman-Enskog Procedure

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- Taking the 1<sup>st</sup> order departure of  $f$  from the Maxwellian distribution as:

$$f(\mathbf{x}, \mathbf{c}, t) = f^{(eq)}(\mathbf{x}, \mathbf{c}, t) + \xi f^{(1)}(\mathbf{x}, \mathbf{c}, t) + \dots$$

- Substituting into the Boltzmann equation:

$$\left(\frac{\partial f}{\partial t}\right)_c = \int \int [f^{(1)'}(\mathbf{x}, \mathbf{c}', t) f^{(eq)'}(\mathbf{x}, \mathbf{z}', t) + f^{(1)'}(\mathbf{x}, \mathbf{z}', t) f^{(eq)'}(\mathbf{x}, \mathbf{c}', t)] |\mathbf{z} - \mathbf{c}| n \sigma(\Omega) dV_z d\Omega \\ - \int \int [f^{(1)}(\mathbf{x}, \mathbf{c}, t) f^{(eq)}(\mathbf{x}, \mathbf{z}, t) + f^{(1)}(\mathbf{x}, \mathbf{z}, t) f^{(eq)}(\mathbf{x}, \mathbf{c}, t)] |\mathbf{z} - \mathbf{c}| n \sigma(\Omega) dV_z d\Omega$$





# Chapman-Enskog Procedure

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- Keeping only 1<sup>st</sup> order terms in the expansion:

$$\left(\frac{\partial f}{\partial t}\right)_c = \int \int [f^{(1)}(\mathbf{x}, \mathbf{c}, t) f^{(eq)'}(\mathbf{x}, \mathbf{z}, t)] |\mathbf{z} - \mathbf{c}| n \sigma(\Omega) dV_z d\Omega \approx -f^{(1)}(\mathbf{x}, \mathbf{c}, t) n \sigma_{tot} \bar{\mathbf{c}}_{rel}$$

- Where

- The average relative velocity btwn particles is  $\bar{\mathbf{c}}_{rel}$
- The total collision cross-section is  $\sigma_{tot}$

- The collision frequency is  $\nu_r = \nu_c = n \sigma_{tot} \bar{\mathbf{c}}_{rel}$



# The “1<sup>st</sup> Order” Boltzmann Equation

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- So the Boltzmann equation becomes, to 1st order:

$$\frac{\partial f}{\partial t} + \mathbf{c} \cdot \nabla f = -\frac{1}{\lambda} (f - f^{(eq)})$$

- Or

$$\frac{df}{dt} + \frac{1}{\lambda} f = \frac{1}{\lambda} f^{(eq)}$$

- On the characteristic line  $\mathbf{c} = d\mathbf{x}/dt$
- Where  $\lambda = 1 / \nu_r$



# Integrating the “1<sup>st</sup> Order” BE

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- Integrating the “1<sup>st</sup> Order” BE over a time step  $\delta_t$ :

$$f(\mathbf{x} + \mathbf{c}\delta_t, \mathbf{c}, t + \delta_t) = \frac{1}{\lambda} e^{-\delta_t/\lambda} \int_0^{\delta_t} e^{t'/\lambda} f^{(eq)}(\mathbf{x} + \mathbf{c}t', \mathbf{c}, t + t') dt' + e^{-\delta_t/\lambda} f(\mathbf{x}, \mathbf{c}, t)$$

- Assuming  $\delta_t$  is small enough &  $f^{(eq)}$  is smooth enough locally, then for  $0 \leq t' \leq \delta_t$ :

$$f^{(eq)}(\mathbf{x} + \mathbf{c}t', \mathbf{c}, t + t') = \left(1 - \frac{t'}{\delta_t}\right) f^{(eq)}(\mathbf{x}, \mathbf{c}, t) + \frac{t'}{\delta_t} f^{(eq)}(\mathbf{x} + \mathbf{c}\delta_t, \mathbf{c}, t + \delta_t) + O(\delta_t^2)$$



# Integrating the “1<sup>st</sup> Order” BE

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- Putting these last 2 eqs. together:

$$f(\mathbf{x} + \mathbf{c}\delta_t, \mathbf{c}, t + \delta_t) - f(\mathbf{x}, \mathbf{c}, t) = (e^{-\delta_t/\lambda} - 1) [f(\mathbf{x}, \mathbf{c}, t) - f^{(eq)}(\mathbf{x}, \mathbf{c}, t)] \\ + \left[ 1 + \frac{\lambda}{\delta_t} (e^{-\delta_t/\lambda} - 1) \right] [f^{(eq)}(\mathbf{x} + \mathbf{c}\delta_t, \mathbf{c}, t + \delta_t) - f^{(eq)}(\mathbf{x}, \mathbf{c}, t)]$$

- Expanding  $e^{-\delta_t/\lambda}$  in a Taylor series while neglecting terms of  $O(\delta_t^2)$  and also defining  $\tau = \lambda/\delta_t$ :

$$f(\mathbf{x} + \mathbf{c}\delta_t, \mathbf{c}, t + \delta_t) - f(\mathbf{x}, \mathbf{c}, t) = -\frac{1}{\tau} [f(\mathbf{x}, \mathbf{c}, t) - f^{(eq)}(\mathbf{x}, \mathbf{c}, t)]$$



# Low Mach Number Approximation

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- In LBE, the equilibrium distribution

$$f^{(eq)} = \left( \frac{m}{2\pi kT} \right)^{D/2} e^{-\frac{m}{2kT}(\mathbf{c}-\mathbf{u})^2} = \left( \frac{m}{2\pi kT} \right)^{D/2} e^{-\frac{m}{2kT}\mathbf{c}^2} e^{-\frac{m}{2kT}(2\mathbf{c}\cdot\mathbf{u}-\mathbf{u}\cdot\mathbf{u})}$$

- is obtained from a truncated small velocity expansion or low-Mach-number approximation

$$f^{(eq)} = \left( \frac{m}{2\pi kT} \right)^{D/2} e^{-\frac{m}{2kT}\mathbf{c}^2} \left[ 1 + \frac{\mathbf{c}\cdot\mathbf{u}}{kT} + \frac{1}{2} \left( \frac{\mathbf{c}\cdot\mathbf{u}}{kT} \right)^2 - \frac{\mathbf{u}\cdot\mathbf{u}}{2kT} \right] + O(u^3)$$

- $D =$  number of dimensions (e.g.,  $D = 3$  for 3D)



# Discretization of Phase Space

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- Discretization of momentum space is coupled to that of configuration space such that a lattice structure is obtained
- This is a special characteristic of LBE
- Quadrature must be accurate enough to
  - Preserve conservation constraints exactly
  - Retain necessary symmetries of Navier-Stokes



# Discretization of Phase Space

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- The first 2 order approximations of the distribution function ( $f^{(eq)}$ ,  $f^{(1)}$ ) are used to derive Navier-Stokes
- So quadrature used must evaluate hydrodynamic moments w.r.t  $f^{(eq)}$  exactly:

$$\rho: \quad 1, c_i, c_i c_j,$$

$$\mathbf{u}: \quad c_i, c_i c_j, c_i c_j c_k,$$

$$T: \quad c_i c_j, c_i c_j c_k, c_i c_j c_k c_l,$$

- Assuming particle has linear d.o.f. only, d.o.f= $D$



# Discretization of Phase Space

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- To obtain Navier-Stokes, must evaluate moments of  $1, \mathbf{c}, \dots, \mathbf{c}^6$ , w.r.t. wt. fnctn  $e^{-m\mathbf{c}\cdot\mathbf{c}/2kT}$  exactly
- Hydro-dynamic moments of  $f^{(eq)}$ :  $I = \int \psi(\mathbf{c}) f^{(eq)} d\mathbf{c}$

$$I = \left( \frac{m}{2\pi kT} \right)^{D/2} \int \psi(\mathbf{c}) e^{-\frac{m}{2kT} \mathbf{c}^2} \left[ 1 + \frac{\mathbf{c} \cdot \mathbf{u}}{kT} + \frac{1}{2} \left( \frac{\mathbf{c} \cdot \mathbf{u}}{kT} \right)^2 - \frac{\mathbf{u} \cdot \mathbf{u}}{2kT} \right] d\mathbf{c}$$

- Use Gaussian-type quadrature to evaluate

$$\int \psi(x) e^{-x^2} dx$$





# Square Lattice-Boltzmann Model 9-bit 2D

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- In Cartesian coordinates:  $\psi_{mn}(\mathbf{c}) = c_x^m c_y^n$
- Then

$$I = \int_{-\infty}^{\infty} \psi_{mn}(\mathbf{c}) f^{(eq)} d\mathbf{c} = \frac{m}{\pi} (\sqrt{2kT})^{m+n} \times \left\{ \left( 1 - \frac{\mathbf{u}^2}{2kT} \right) I_m I_n + \frac{2(u_x I_{m+1} I_n + u_y I_m I_{n+1})}{\sqrt{2kT}} + \frac{u_x^2 I_{m+2} I_n + 2u_x u_y I_{m+1} I_{n+1} + u_y^2 I_m I_{n+2}}{RT} \right\}$$

- Where

$$I_m = \int_{-\infty}^{\infty} e^{-\zeta^2} \zeta^m d\zeta, \quad \zeta = c/\sqrt{2kT}$$



# Square Lattice-Boltzmann Model 9-bit 2D

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- Use 3rd order Hermite formula to evaluate

$$I_m = \int_{-\infty}^{\infty} e^{-\zeta^2} \zeta^m d\zeta = \sum_{j=1}^3 \omega_j \zeta_j^m$$

- Where the 3 abscissas of the quadrature are:

$$\zeta_1 = -\sqrt{3/2}, \zeta_2 = 0, \zeta_3 = \sqrt{3/2}$$

- And the corresponding weight coefficients are:

$$\omega_1 = \sqrt{\pi} / 6, \omega_2 = 2\sqrt{\pi} / 3, \omega_3 = \sqrt{\pi} / 6$$



# 2D Square Lattice-Boltzmann Model 9-bit

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- Moment integral becomes

$$I_m = \frac{m}{\pi} \sum_{i,j=1}^3 \omega_i \omega_j \psi(c_{i,j}) \left\{ 1 + \frac{c_{i,j} \bullet \mathbf{u}}{kT} + \frac{(c_{i,j} \bullet \mathbf{u})^2}{2(kT)^2} - \frac{\mathbf{u}^2}{2kT} \right\}$$

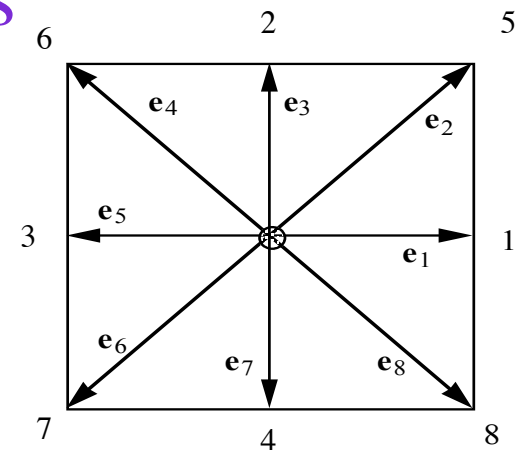
- From which, parts of the equilibrium distribution function are identified as

$$f_{i,j}^{(eq)} = \frac{m}{\pi} \omega_i \omega_j \left\{ 1 + \frac{c_{i,j} \bullet \mathbf{u}}{kT} + \frac{(c_{i,j} \bullet \mathbf{u})^2}{2(kT)^2} - \frac{\mathbf{u}^2}{2kT} \right\}$$



# 2D Square Lattice-Boltzmann Model 9-bit

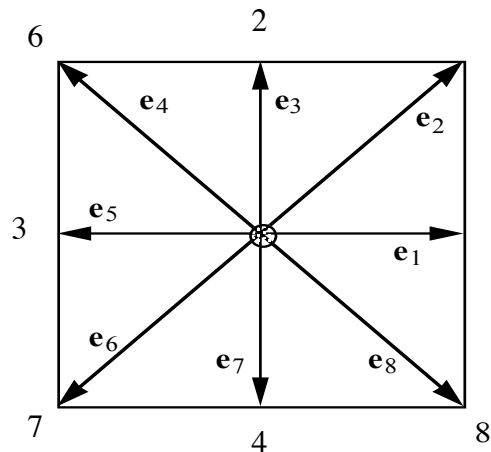
- Select possible molecular velocities on 2D square lattice to go in as many directions of such square
- The chosen velocities have a certain symmetry to account for molecules moving in any and all directions independent of directions (isotropy)
- For square this means sides & corners
- This means 9 possible velocities
- To go w/at least 9 terms in  $\psi_{mn}(\mathbf{c})$



# 2D Square Lattice-Boltzmann Model 9-bit

- The 9 possible velocities are:

$$e_{\alpha} = \begin{cases} (0,0), & \alpha = 0, \\ (\cos\theta_{\alpha}, \sin\theta_{\alpha})c, & \theta_{\alpha} = (\alpha - 1)\pi / 2, \quad \alpha = 1,2,3,4, \\ \sqrt{2}(\cos\theta_{\alpha}, \sin\theta_{\alpha})c, & \theta_{\alpha} = (\alpha - 5)\pi / 2 + \pi / 4, \quad \alpha = 5,6,7,8 \end{cases}$$



# 2D Square Lattice-Boltzmann Model 9-bit

- Then parts of the equilibrium distribution function are identified as

$$f_{\alpha}^{(eq)} = mw_{\alpha} \left\{ 1 + \frac{\mathbf{e}_{\alpha} \bullet \mathbf{u}}{c^2} + \frac{(\mathbf{e}_{\alpha} \bullet \mathbf{u})^2}{2c^4} - \frac{\mathbf{u}^2}{2c^2} \right\}$$

- Where the corresponding weight coefficients are now

$$w_{\alpha} = \frac{\omega_i \omega_j}{\pi} = \begin{cases} 4/9, & i = j = 2, & \alpha = 0, \\ 1/9, & i = 1, j = 2, \dots, & \alpha = 1, 2, 3, 4, \\ 1/36, & i = j = 1, & \alpha = 5, 6, 7, 8 \end{cases}$$

- And  $RT = c_s^2 = c^2/3$  or

$$\left\| \sqrt{RT} \vec{\zeta} \right\| = \sqrt{3RT} = c$$



# 3D Cube Lattice-Boltzmann Model 27-bit

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- This is a straight-forward extension of 2D:

$$I_m = \frac{m}{\pi^{3/2}} \sum_{i,j,k=1}^3 \omega_i \omega_j \omega_k \psi(c_{i,j,k}) \left\{ 1 + \frac{c_{i,j,k} \bullet \mathbf{u}}{kT} + \frac{(c_{i,j,k} \bullet \mathbf{u})^2}{2(kT)^2} - \frac{\mathbf{u}^2}{2kT} \right\}$$

- where

$$e_\alpha = \begin{cases} (0,0,0) & \alpha = 0, \\ (\pm 1,0,0)c, (0,\pm 1,0)c, (0,0,\pm 1)c, & \alpha = 1,2,\dots,6, \\ (\pm 1,\pm 1,0)c, (\pm 1,0,\pm 1)c, (0,\pm 1,\pm 1)c, & \alpha = 7,8,\dots,18, \\ (\pm 1,\pm 1,\pm 1)c, & \alpha = 19,20,\dots,26 \end{cases}$$



# 3D Cube Lattice-Boltzmann Model 27-bit

---

- Then parts of the equilibrium distribution function are identified as

$$f_{\alpha}^{(eq)} = mw_{\alpha} \left\{ 1 + \frac{3\mathbf{e}_{\alpha} \bullet \mathbf{u}}{c^2} + \frac{9(\mathbf{e}_{\alpha} \bullet \mathbf{u})^2}{2c^4} - \frac{3\mathbf{u}^2}{2c^2} \right\}$$

- where

$$w_{\alpha} = \begin{cases} 8/27 & i = j = k = 2 & \alpha = 0, \\ 2/27, & i = j = 2, k = 1, \dots, & \alpha = 1, 2, \dots, 6, \\ 1/54, & i = j = 1, k = 2, \dots, & \alpha = 7, 8, \dots, 18, \\ 1/216, & i = j = k = 1, \dots, & \alpha = 19, 20, \dots, 26 \end{cases}$$





# Lattice Boltzmann Equation

- Equilibrium distribution function for  $f_\alpha$

$$f_\alpha^{(eq)} = \rho w_\alpha \left[ 1 + \frac{3}{c^2} \mathbf{e}_\alpha \cdot \mathbf{u} + \frac{9}{2c^4} (\mathbf{e}_\alpha \cdot \mathbf{u})^2 - \frac{3}{2c^2} \mathbf{u} \cdot \mathbf{u} \right]$$

Δ 9-velocity model(2D):

$$w_\alpha = \begin{cases} 4/9, & \alpha = 0 \\ 1/9, & \alpha = 1,3,5,7 \\ 1/36, & \alpha = 2,4,6,8 \end{cases}$$

Δ 15-velocity model:

$$w_\alpha = \begin{cases} 2/9, & \alpha = 0 \\ 1/9, & \alpha = 1,2,\dots,6 \\ 1/72, & \alpha = 7,8,\dots,14. \end{cases}$$

Δ 19-velocity model:

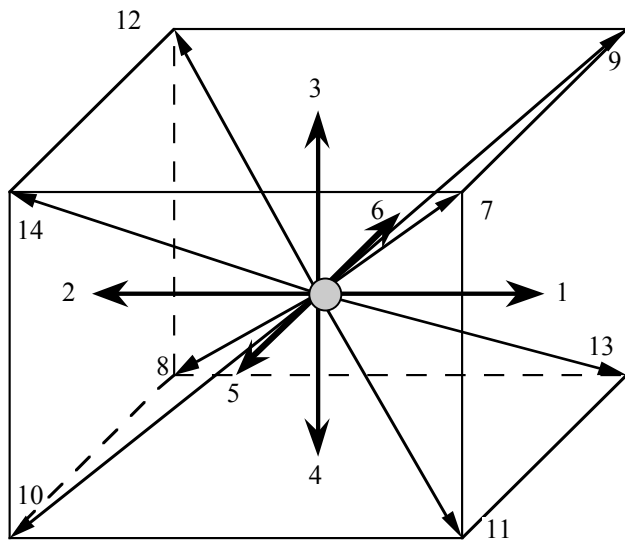
$$w_\alpha = \begin{cases} 1/3, & \alpha = 0 \\ 1/18, & \alpha = 1,2,\dots,6 \\ 1/36, & \alpha = 7,8,\dots,18. \end{cases}$$

Qian YH, d'Humieres D, Lallemand P. Lattice BGK Models for Navier Stokes Equation. Europhys Lett 1992;17:479-484.



# Examples of 3-D lattice models

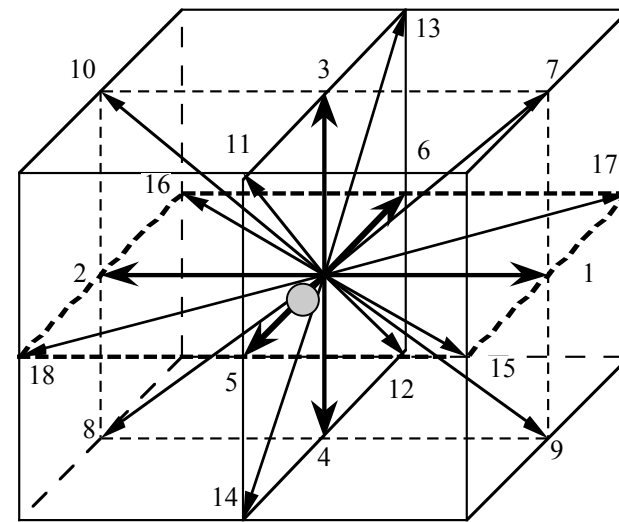
15-velocity **3-D**  
lattice model



Q15D3 lattice

←  $2\delta x$  →

19-velocity **3-D**  
lattice model



Q19D3 lattice

←  $2\delta x$  →

# Lattice Boltzmann Equation

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- Macroscopic variables is obtained from:

$$\rho = \sum_{\alpha=0}^N f_{\alpha} = \sum_{\alpha=0}^N f_{\alpha}^{(\text{eq})}, \quad \rho \mathbf{u} = \sum_{\alpha=1}^N \mathbf{e}_{\alpha} f_{\alpha} = \sum_{\alpha=1}^N \mathbf{e}_{\alpha} f_{\alpha}^{(\text{eq})}$$

- Chapman-Enskog analysis (multi-scale expansion)  $\Rightarrow$   
NS Eqs. recovered in near incompressible flow limit.
- Equation of state:  $p = \rho c_s^2 = \rho / 3$
- What are the **advantages**?
- What are potential **benefits** comparing with the standard CFD methods for the Navier-Stokes equations?
- Let's look at the actual implementation:

LBGK scheme.



# Lattice Boltzmann Equation

**LBGK Scheme:** discretization in time & space  $\Rightarrow$

$$f_a(\mathbf{x}_i + \mathbf{e}_a dt, t + dt) - f_a(\mathbf{x}_i, t) = -\frac{1}{\tau} [f_\alpha(\mathbf{x}_i, t) - f_\alpha^{(eq)}(\mathbf{x}_i, t)]$$

- Viscosity:  $\nu = (\tau - 1/2) c_s^2 \delta t$
- Order of accuracy: 2nd in  $\mathbf{x}$  & 1st in  $t$ .
- Computation:
  - collision step:  $\tilde{f}_\alpha(\mathbf{x}_i, t) - f_\alpha(\mathbf{x}_i, t) = -\frac{1}{\tau} [f_\alpha(\mathbf{x}_i, t) - f_\alpha^{(eq)}(\mathbf{x}_i, t)]$
  - streaming step:  $f_a(\mathbf{x}_i + \mathbf{e}_a dt, t + dt) = \tilde{f}_\alpha(\mathbf{x}_i, t)$
- Advantages:
  - collision step is local; streaming step takes no computation.
  - explicit in form, easy to implement, and natural to parallelize.
  - Pressure is obtained simply as:  $p = \rho c_s^2 = \rho / 3$



# Computational Procedure

Initialization:

Assign  $f^{(eq)}$   $f_\alpha = f_\alpha^{(eq)} = \rho w_\alpha \left[ 1 + \frac{3}{c^2} \mathbf{e}_\alpha \cdot \mathbf{u} + \frac{9}{2c^4} (\mathbf{e}_\alpha \cdot \mathbf{u})^2 - \frac{3}{2c^2} \mathbf{u} \cdot \mathbf{u} \right]$

Collision:

$$\tilde{f}_\alpha^{(t)} = f_\alpha^{(t)} - \frac{1}{\tau} \left[ f_\alpha^{(t)} - f_\alpha^{(t,eq)} \right]$$

$t = t + \delta t$

Streaming:

$$f_\alpha^{(t+\delta t)}(x_i + e_i \delta t, t + \delta t) = f_\alpha^{(t)}(x_i, t)$$

Calculate physical variables

$$\rho^{(t+\delta t)} = \sum_{\alpha=0}^N f_\alpha^{(t+\delta t)} \quad \rho^{(t+\delta t)} \mathbf{u}^{(t+\delta t)} = \sum_{\alpha=0}^N \mathbf{e}_\alpha f_\alpha^{(t+\delta t)}$$



# Boltzmann's $H$ -theorem

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- Generally, macroscopic processes are irreversible.
- The relaxation to a Maxwellian distribution as a result of collisions, is an irreversible process.
- $H$ -theorem states that if the distribution function evolves according to the Boltzmann equation, then for a uniform gas in the absence of external forces  $H$  can never increase:  
$$\frac{dH}{dt} < 0 \qquad H = \int_{-\infty}^{\infty} f \log f dV_c$$
- if we begin with a uniform gas having a non-equilibrium distribution function,  $H$  decreases until the gas relaxes to the equilibrium distribution when  $H$  attains a minimum value



# Applications of LBE

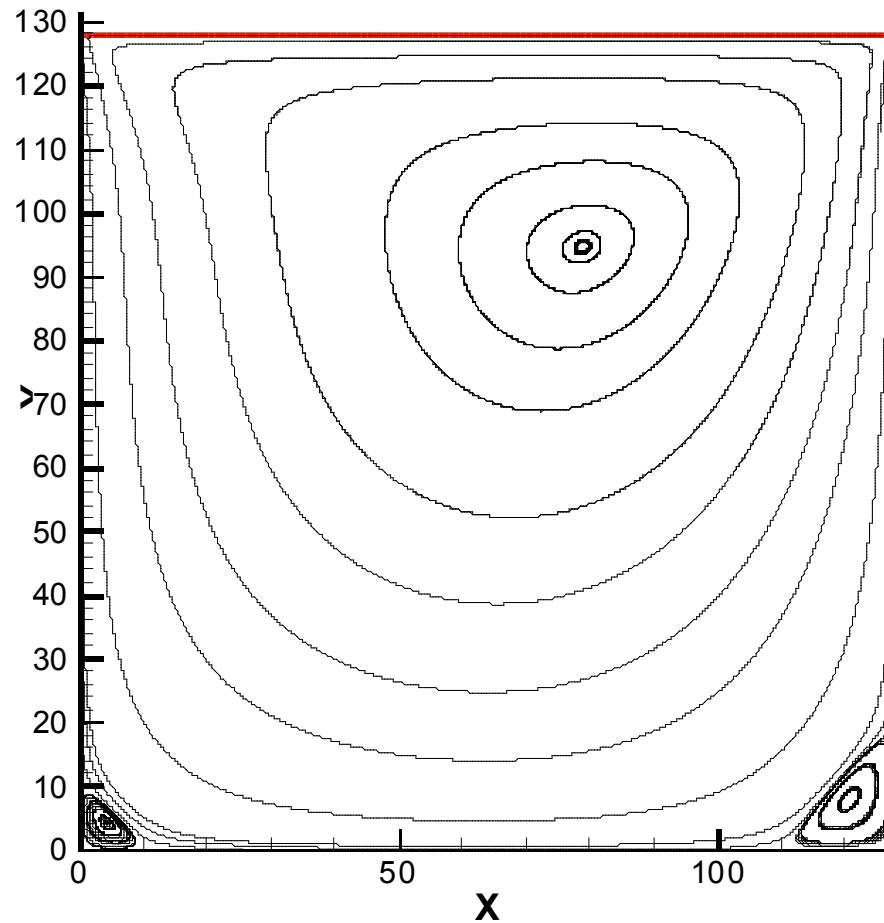
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- Simulation of incompressible flows
- Fully compressible and thermal flows
- Multi-phase and multi-component flows
- Particulate Suspensions
- Turbulent Flows
- Micro Flows



# Streamlines in the cavity flow at $Re=100$

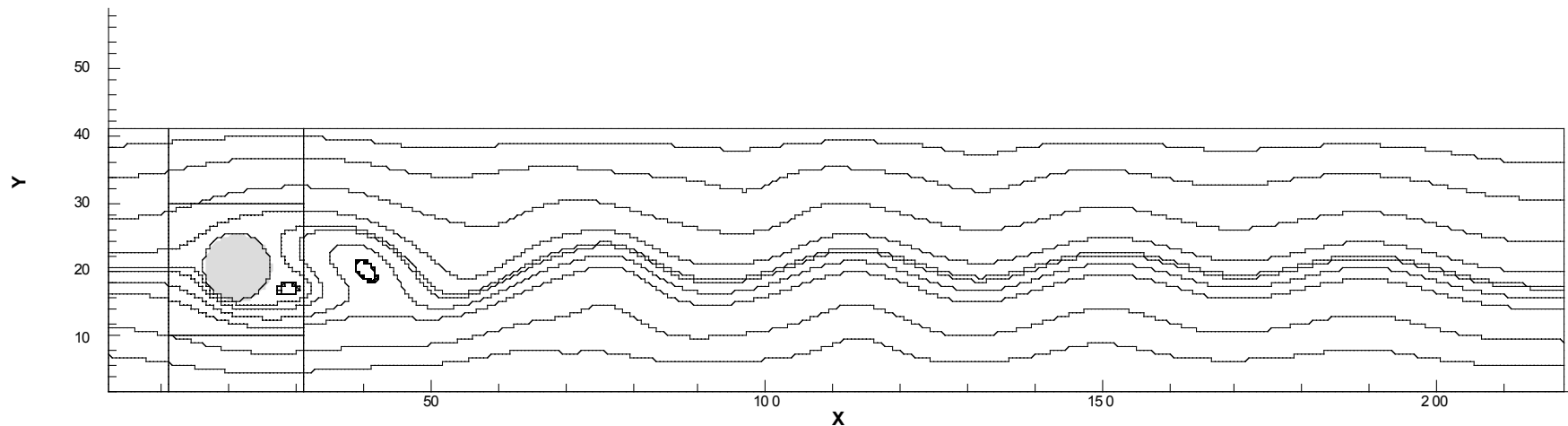
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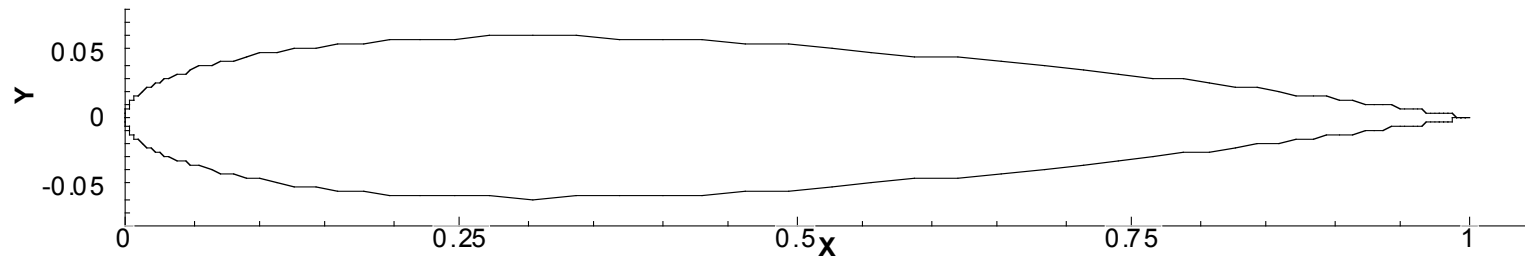
# Instantaneous streamlines for channel flow over an asymmetrically placed cylinder at $Re=100$

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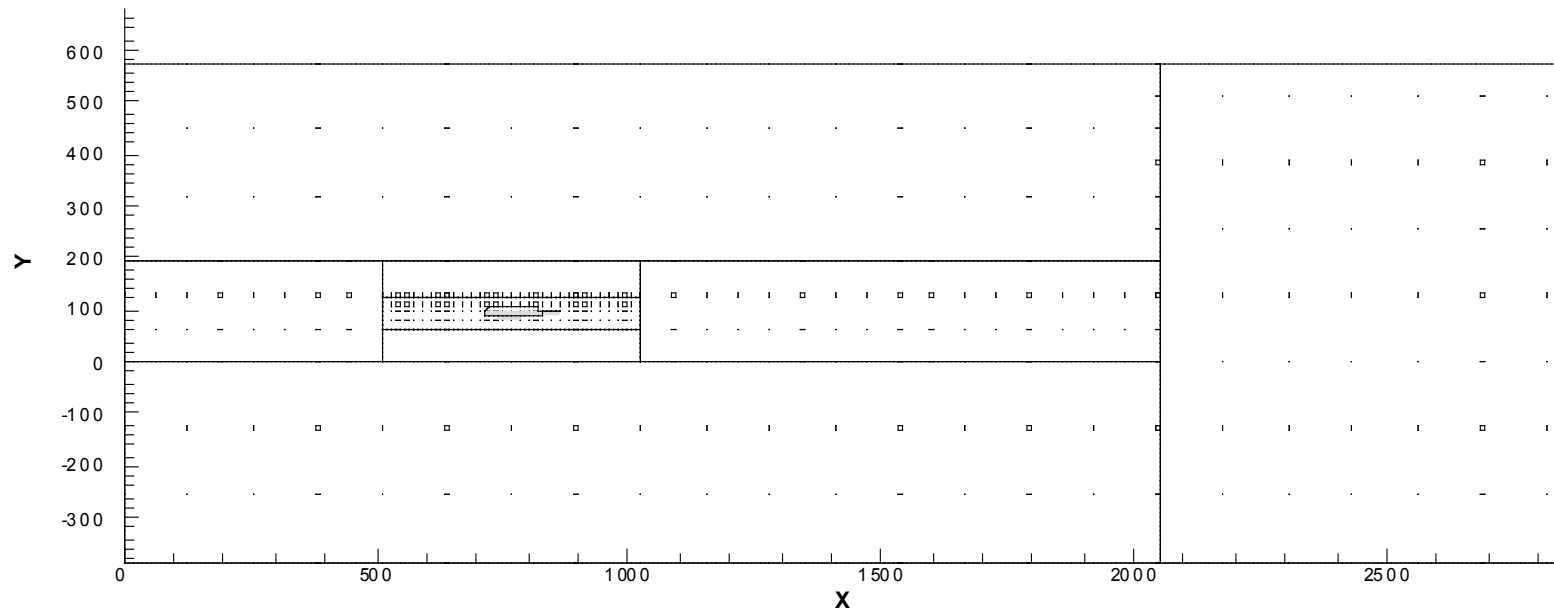
# NACA 0012 airfoil

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# Block and lattice layout for flow over NACA 0012

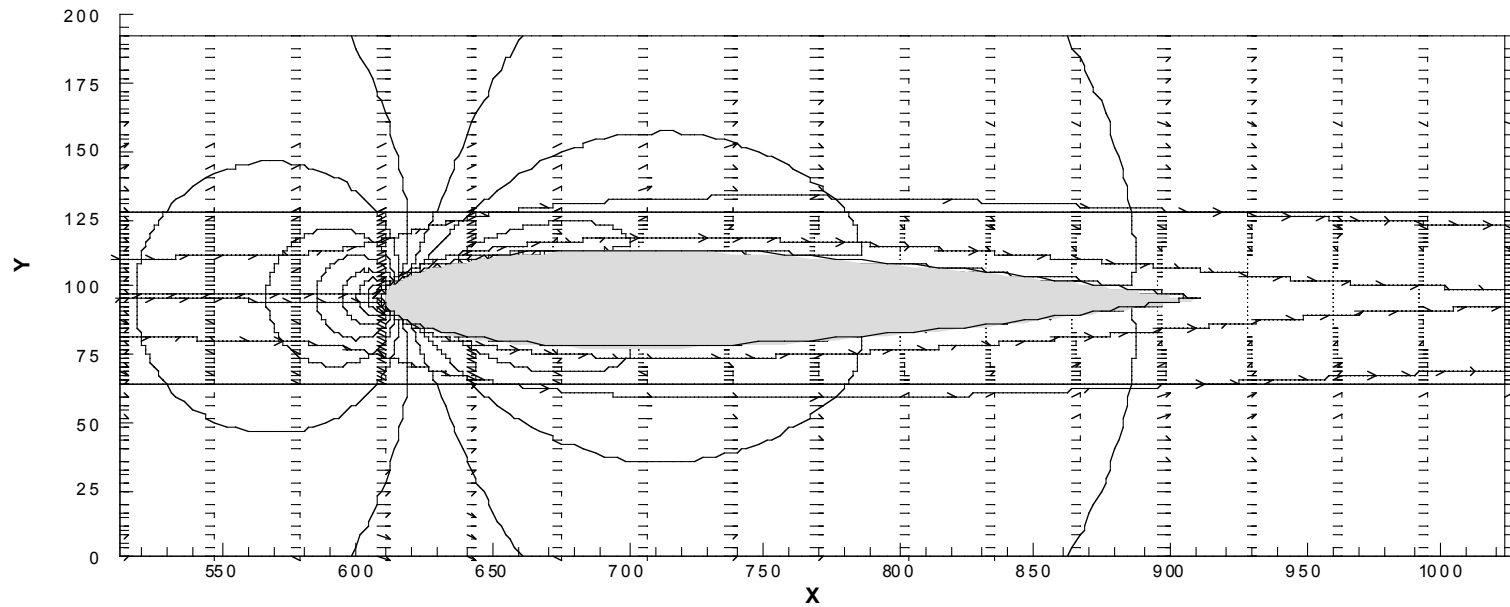
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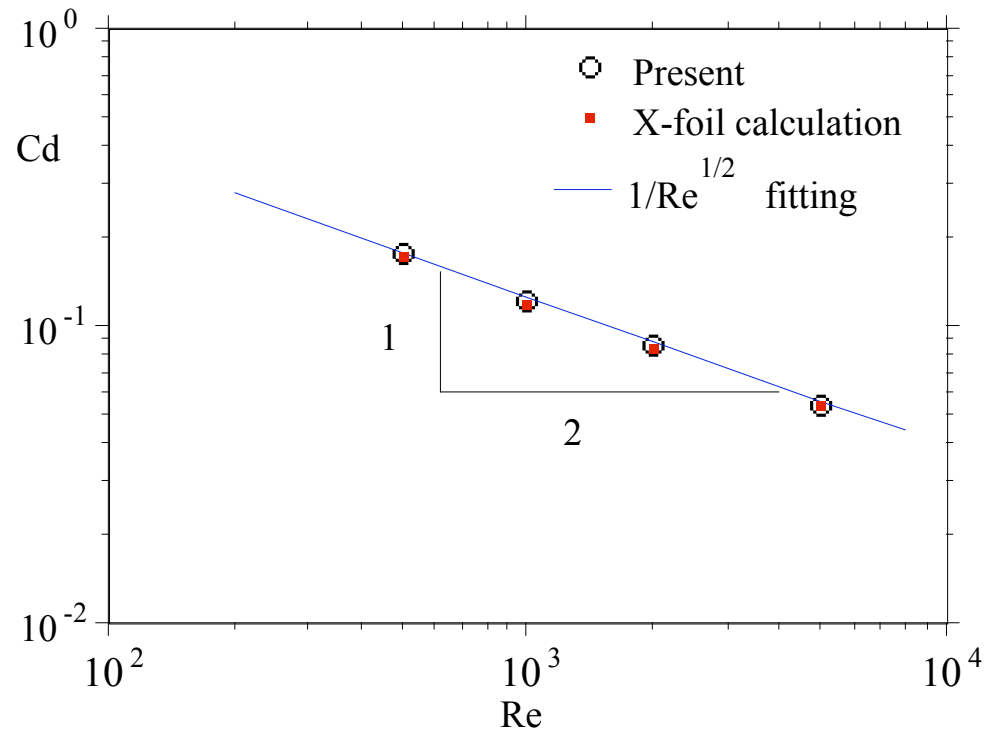
- The lattice spacing is reduced by a factor 32 for graphical clarity



# Streamlines, pressure contour, velocity vector for uniform flow over NACA 0012 airfoil at $Re=2000$ .



# Comparing $C_d$ between present simulation and Xfoil calculation vs. $Re$ for NACA0012 flow.



- The straight line is the slope according to the laminar boundary layer theory

# Rayleigh-Taylor Instability

Single mode perturbation

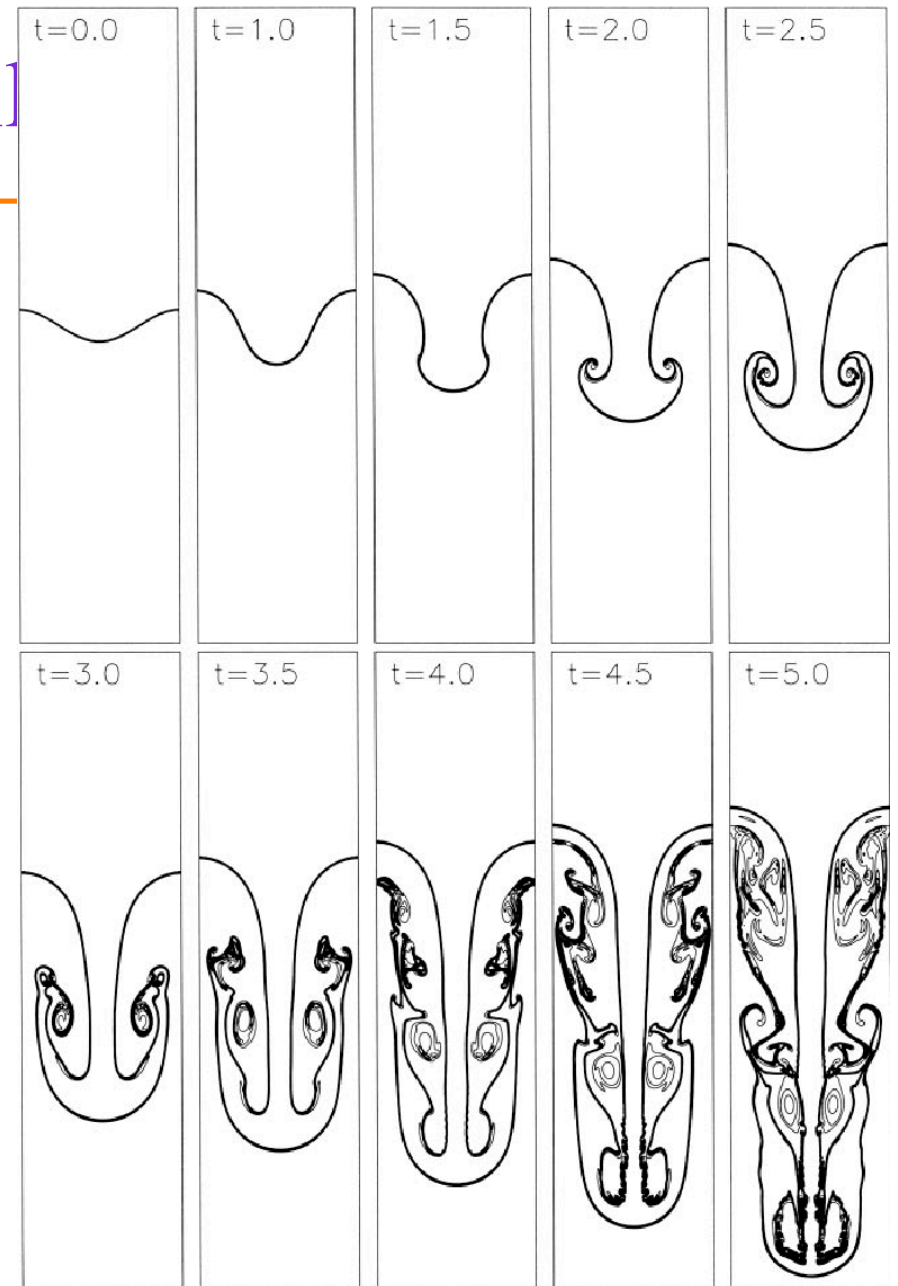
$W$  = channel width

$\rho_h$  = heavy fluid

$\rho_l$  = lighter fluid

$$\text{Re} = \frac{W \sqrt{Wg}}{\nu} = 2048$$

$$A = \frac{\rho_h - \rho_l}{\rho_h + \rho_l} = 0.5$$



# Implementation of complex LBE model.

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- For complicated problems, new LBE models may need to be designed.
- The number of particle velocities in new models can vary.
- Hybrid modes which incorporate finite different method in LBE need to be considered.



# References for LBE

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Chen H, Chen S, Matthaeus WH. Recovery of the Navier-Stokes equations using a lattice-gas Boltzmann method. Phys Rev A 1992;45:R5339-R5342.

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Bhatnagar PL, Gross EP, Krook M. A model for collision processes in gases, I. small amplitude processes in charged and neutral one-component system. Phys Rev 1954;94:511-525.

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He X, Luo L-S. Theory of the lattice Boltzmann equation: From Boltzmann equation to lattice Boltzmann equation. Phys Rev E 1997;56:6811-6817.





# Boltzmann Equation

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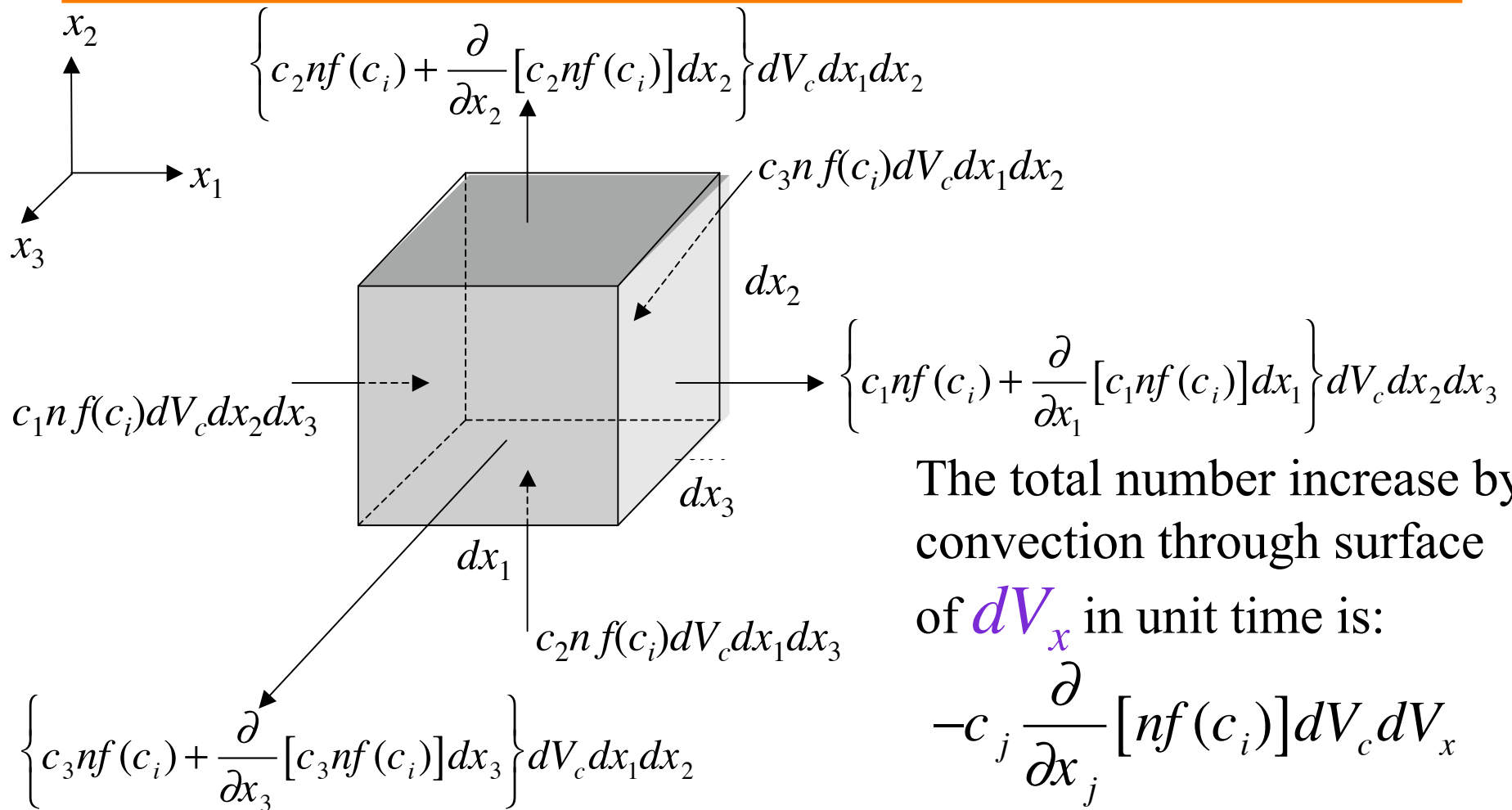
- Studying molecules (class  $c_i$ ) inside  $dV_x$ , we see total number of molecules whose velocity is between  $c_i$  and  $c_i + dV_c$  is

$$nf(c_i) dV_x dV_c$$

- Change of number of molecules in class  $c_i$  must result from convection of molecules across the surface of  $dV_c$  and  $dV_x$  or from intermolecular collision within



# Number increased by convection thru $dV_x$



The total number increase by convection through surface of  $dV_x$  in unit time is:

$$-c_j \frac{\partial}{\partial x_j} [n f(c_i)] dV_c dV_x$$

$c_i$  is taken out of the differentiation because of it is independent of  $x_j$

# Boltzmann Equation

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- If there is no external force, the convection of molecules through the surface of  $dV_c$  is zero.
- The rate of increase of number of molecules in class  $c_i$  results from collision is:

$$\left\{ \frac{\partial}{\partial t} [nf(c_i)] \right\}_{collision} dV_c dV_x$$

- The total rate of increase of number of molecules of class  $c_i$  is:

$$\frac{\partial}{\partial t} [nf(c_i)] dV_c dV_x$$



# Boltzmann Equation

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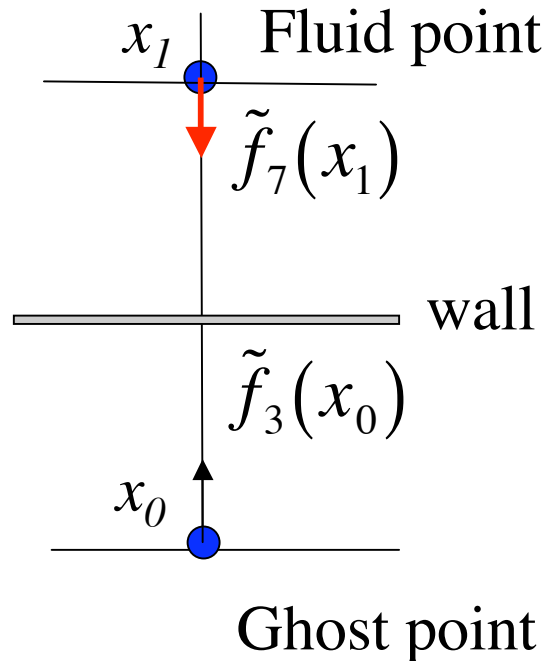
- The rate of increase of the number of molecules of class  $c_i$  in the volume element is equal to the rate of increase by convection through surface of  $dV_x$  plus the rate of increase by collision.
- This gives the Boltzmann Equation:

$$\frac{\partial}{\partial t} [nf(c_i)] + c_j \frac{\partial}{\partial x_j} [nf(c_i)] = \left\{ \frac{\partial}{\partial t} [nf(c_i)] \right\}_{collision}$$

- How to determine the collision operator?



# Solid wall boundary



After collision, set:

$$\tilde{f}_3(x_0) = \tilde{f}_7(x_1)$$

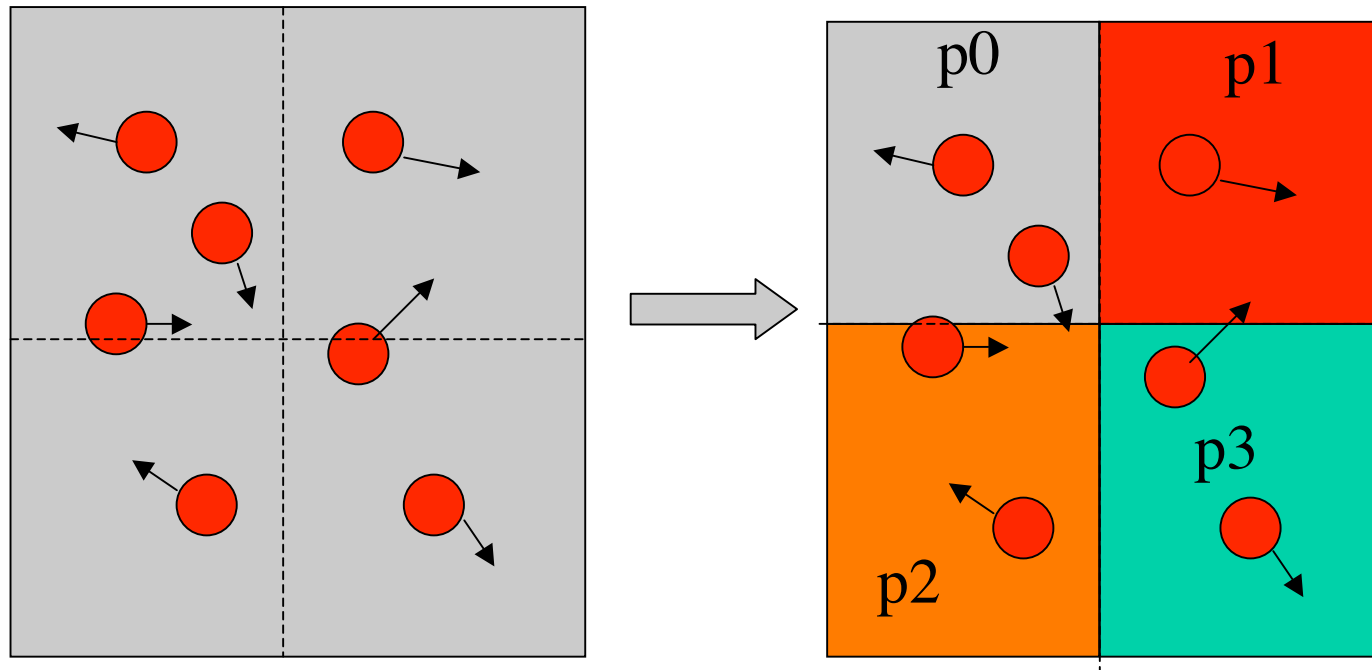
bounce back idea

If the wall has a velocity  $\mathbf{u}$ , momentum flux can be added:

$$\tilde{f}_3(x_0) = \tilde{f}_7(x_1) + 6\rho w_3(\mathbf{e}_3 \cdot \mathbf{u})$$

# Implementation in moving boundary problems

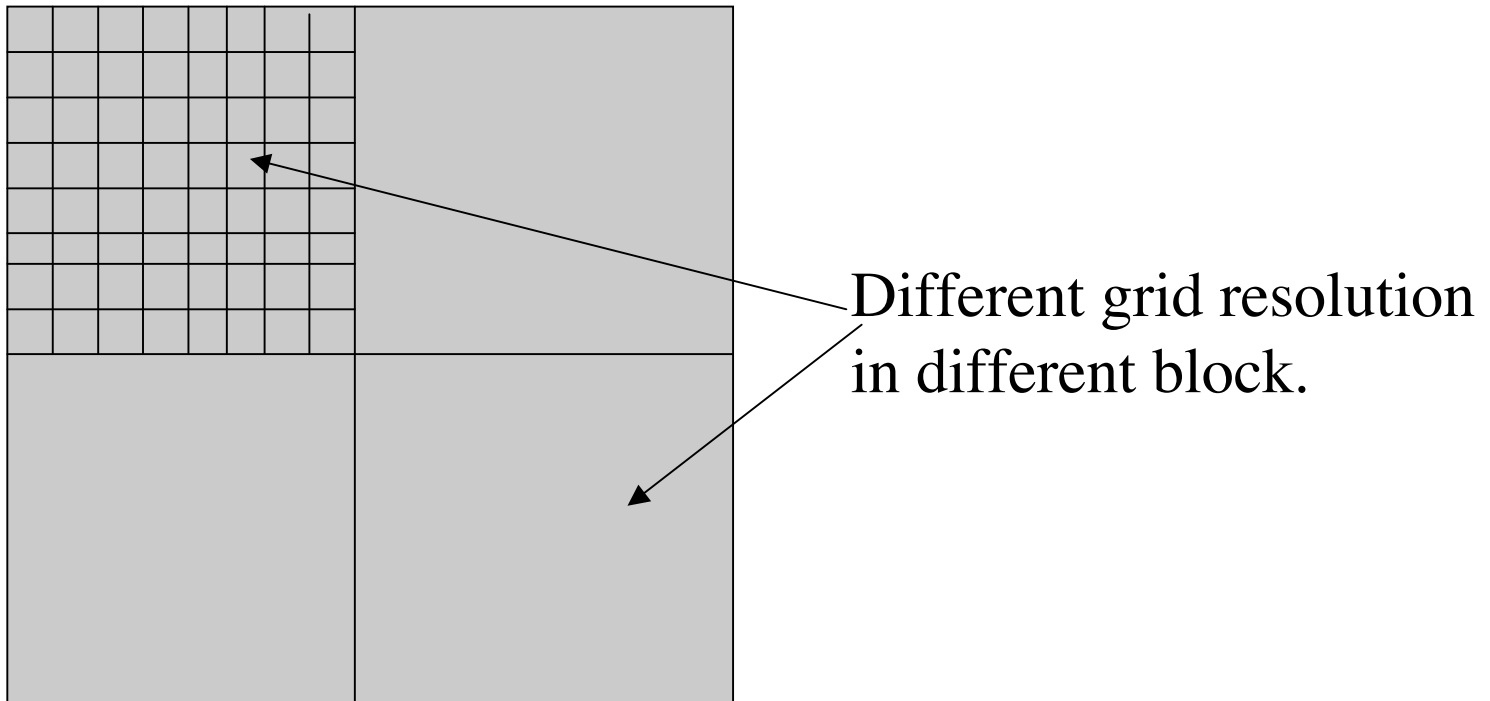
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Particles are moving; they may belong to different processors at different time.

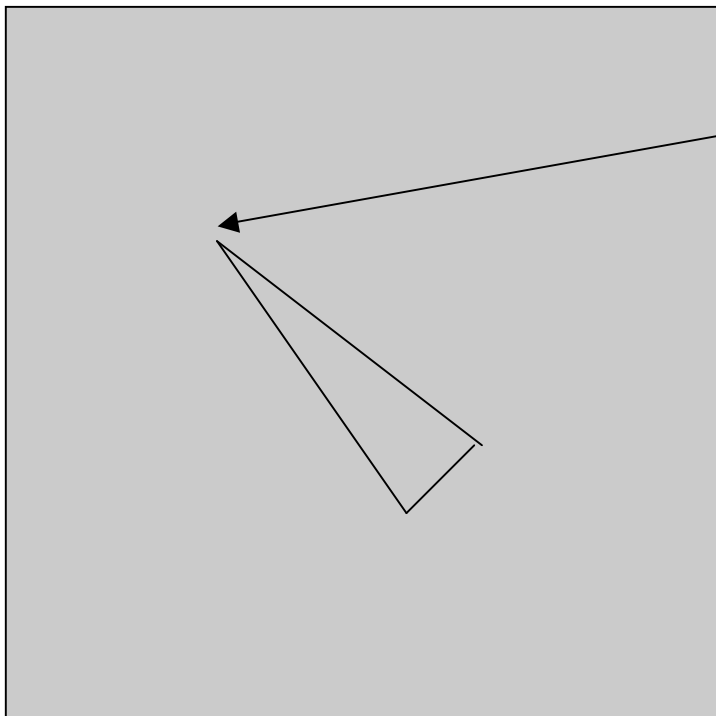
# Multi-block method in LBE

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# Implementation of adaptive grid method in LBE

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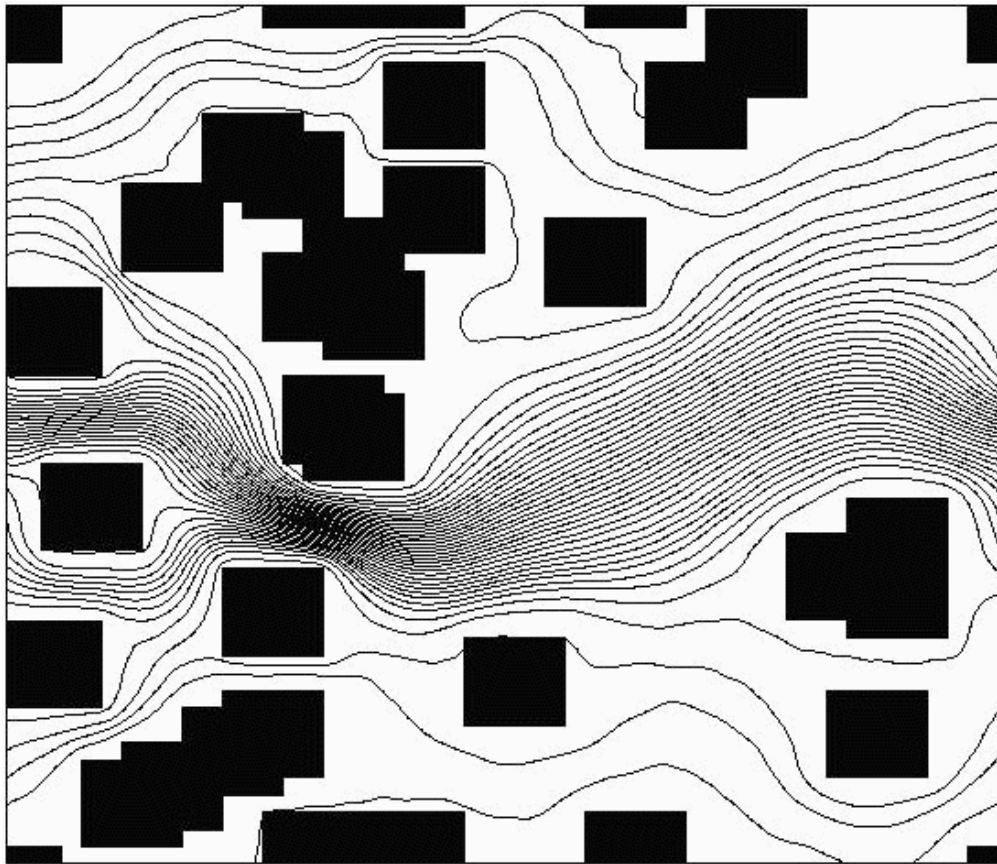


Around the corner, we put fine grids.  
We may also need to increase grid resolution during computation.



# Flows in porous media

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In solid region,  
computation and  
memory are not  
required.

# Boltzmann's $H$ Theorem

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- Generally, entropy,  $S$ , is defined to be related to the # of possible arrangements of molecules
- In other words, entropy is a function of # of possible micro-states,  $\Omega$
- The probability distribution function provides a way of determining the # of possible macro-states,
- Because  $\Omega$  for combined systems of certain micro-states is a product of each,  $\Omega_{AB} = \Omega_A \Omega_B$
- And because total entropy is  $S_{AB} = S_A + S_B$

$$S = -k \log \Omega = -k \int f \log f dV_c,$$



# Boltzmann's $H$ Theorem

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- Differentiating  $H$ , using Boltzmann's eq., & combining terms:

$$\frac{dH}{dt} = -\frac{1}{4} \iint [f(\mathbf{c}')f(\mathbf{z}') - f(\mathbf{c})f(\mathbf{z})] \log \frac{f(\mathbf{c})f(\mathbf{z})}{f(\mathbf{c}')f(\mathbf{z}')} |\mathbf{c} - \mathbf{z}| \sigma d\Omega dV_z \leq 0$$

- $H$  theorem shows why entropy is automatically satisfied if Boltzmann's eq. is satisfied
- Many CFD schemes do not easily guarantee that they will not violate entropy
- Through solving BE,
  - entropy is inherently & automatically satisfied and
  - fluid field is found from  $f$  rather than using  $f$  to find properties to use in Navier-Stokes eqs. derived from BE

